A QED-Compatible Wave Theory of Light, Electrons, and their Interactions

Henry Lindner www.henrylindner.net

Waves vs. Particles

- Wave Theory—Light a wave in an electromagnetic (EM) medium
- Particle Theory—Light is a particle flying through a void

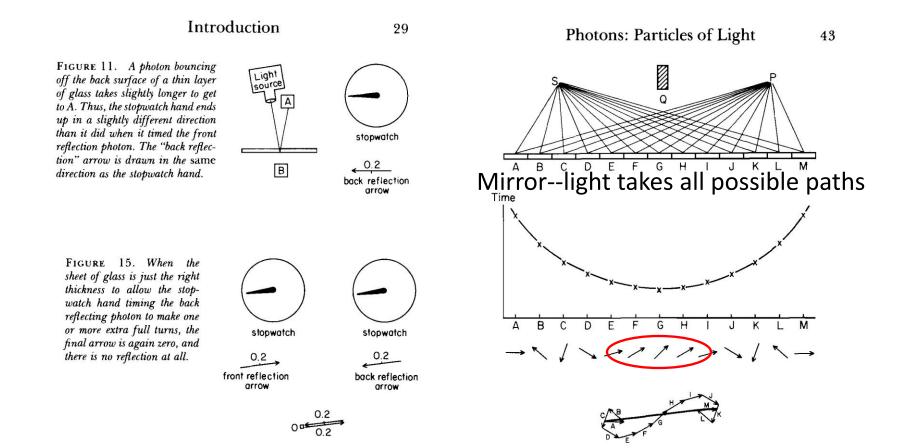
- Different physical qualities and implications.
- Both theories cannot be true of freely propagating light.
- It should be easy to tell with theory is true.
- Wave-particle duality is a contradiction; a violation of the first rule of philosophy.

| Explains or can accommodate: | Wave in Medium | | Particle in Void | |
|---|----------------|-------------|------------------|-----------|
| Wavelength and frequency | Yes | | No | |
| Polarization | Yes | | No | |
| Invariant velocity independ. of source velocity | Yes | "Classical" | No | Falsified |
| Superpositioning (interference) | Yes | Waves | No | Theory |
| Spreading, Diffraction | Yes | | No | |
| Continuous spectrum (gamma to radio waves) | Yes | | No | |
| Laser | ? | | ? | |
| Blackbody Effect | ? | | ? | |
| Photoelectric Effect | ? | | ? | |
| Compton Effect | ? | | ? | |
| Anti-correlation and other photon experiments | ? | | ? | |
| Quantum Electrodynamics | ? | | ? | |

Wave-Particle Truth Table

Quantum Electrodynamics Works

- In QED, Richard Feynman presents QED as a mathematical scheme to predict observations.
- QED is **not** a physical theory of light and electrons.
- In QED, light sources produce amplitude vector arrows that shrink with distance (inverse square law) and rotate with time (accord. to frequency): "shrinks and turns"
- These amplitudes "go everywhere" at c. (Huygens-Fresnel wave model)
- Adding up the resultant arrows for all possible paths renders a final amplitude.
- The square of the amplitude represents the probability of an observed lightmatter interaction occurring at that place and time.



Amplitude arrows "shrink" with distance and "turn" with frequency in time.

Where amplitudes add up due to similar paths (least time) is "where the photon goes".

FIGURE 24. Each path the light could go (in this simplified situation) is shown at the top, with a point on the graph below it showing the time it takes a photon to go from the source to that point on the mirror, and then to the photomultiplier. Below the graph is the direction of each arrow, and at the bottom is the result of adding all the arrows. It is evident that the major contribution to the final arrow's length is made by arrows E through I, whose directions are nearly the same because the timing of their paths is nearly the same. This also happens to be where the total time is least. It is therefore approximately right to say that light goes where the time is least.

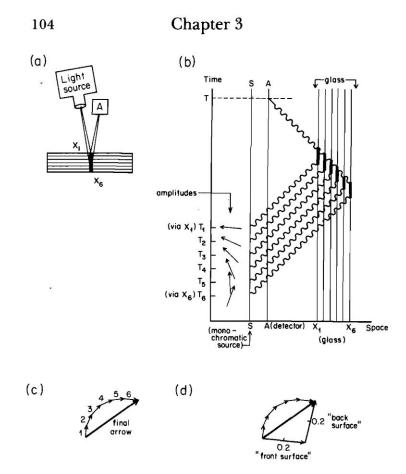
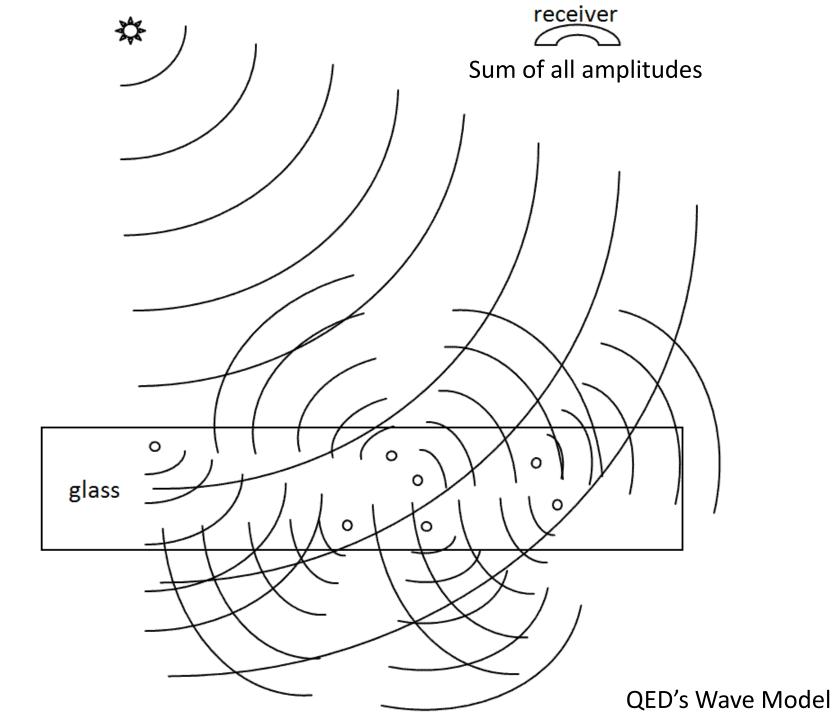


FIGURE 68. We begin our new analysis of partial reflection by dividing a layer of plass into a number of sections (here. six). and looking at the various

Beyond the Classical Model of Reflection

Light amplitudes are <u>absorbed</u> and then <u>re-emitted</u> (scattered in all directions) by electrons **throughout** the glass



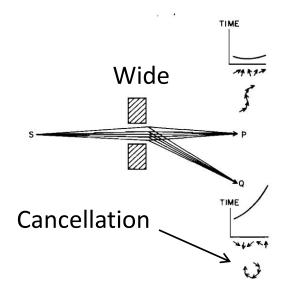


FIGURE 33. Light travels in not just the straight-line path, but in the nearby paths as well. When two blocks are separated enough to allow for these nearby paths, the photons proceed normally to P, and hardly ever go to Q.

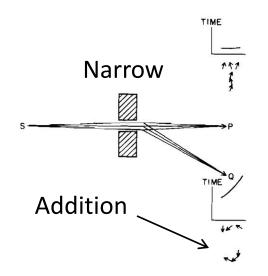


FIGURE 34. When light is restricted so much that only a few paths are possible, the light that is able to get through the narrow slit goes to Q almost as much as to P, because there are not enough arrows representing the paths to Q to cancel each other out.

Diffraction:

Wide aperture: More paths to Q of different lengths, **arrows cancel out Narrow aperture:** Fewer paths to Q of similar length, **arrows add up**

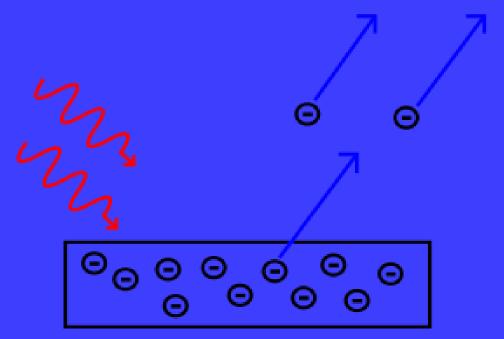
Light amplitudes go everywhere, we only "see" light where they add up! Wave spreading/superpositioning restated as a probability model!

QED is a Wave Model of Light

- For all waves, amplitude² (ψ^2) equals intensity—the rate of flow of wave energy of that frequency!
- In QED, where the wave-energy is most intense is where electrons are most likely to absorb a quantum – where the "photon" is likely to "go".
- Feynman admits that there is no actual photon flying from A to B!
- He knows that trying to say which way the photon went produces paradoxes, yet he can't prevent himself from thinking about flying photons!
- Feynman somehow still believes that "light is made of particles" and so concludes that "Nature is absurd!"
- Feynman admits that all aspects of light and light-matter interactions are explicable by wave theory except the photoelectric effect.

The Photoelectric Effect

Incident UV or x-radiation frees electrons from metal



Quantized Light-matter interactions Low-energy phenomena -- Photoelectric effect Mid-energy phenomena -- Compton scattering High-energy phenomena -- Pair production

Arguments Against Wave Theory

- Frequency Dependence and Frequency Cut Off: According to classical waveparticle interactions, more intense radiation of any frequency should produce higher-energy electrons. More intense EM wave-energy of any frequency should cause some electrons to be ejected.
- Findings: The kinetic energy of the ejected electrons depends only on the light frequency, not on the intensity. No electrons are ejected when the frequency is below the cut-off frequency, no matter how intense the radiation.
- **Conclusion:** Light is made of flying particles (?)

Eisberg, R., Resnick, R., Quantum Physics, 33 (1974)

- Invalid Argument: Findings prove only that classical wave-particle interactions do not explain these quantized interactions.
- Reasonable Conclusion: Electrons absorb EM wave-energy in quanta whose energy is determined only by frequency-wavelength.

Lamb, W., Scully, M., "The photoelectric effect without photons" (1969)

No Time Lag

- According to classical wave and particle theory, wave-energy emitted by a single electron should be distributed equally over a spherically-spreading wavefront, therefore the receiving electron should require time to absorb enough energy to be ejected.
- Finding: No such time lag is observed—all the energy from a presumed single electron emission is absorbed by a nearby electron nearly instantaneously.
- Conclusion: Light is made of flying particles (?)
- Invalid Argument: Evidence only proves that the classical wave-particle model is inadequate to explain the phenomenon.
- Needed: A non-classical wave theory of light and electrons and their interactions that incorporates QED!

Wave Theory of Light and Electrons

- An Electron is an Extended EM Wave-Structure: It is not a point particle. It is composed of circulating EM waves. It is not a particle associated with a field; it is its EM field. It is as large as its influence in space.
- 2. The Electronic Wave-Structure is Quantized: The amplitude and other physical parameters of an electron's EM waves are fixed by its structure. Only the wavelength-frequency varies and determines the momentum of a free electron (de Broglie relation: $\rho = h/\lambda$).
- Electrons Incorporate and Expel EM Waves: Absorbed waves increase the electron's frequency and therefore its wave-energy. When electrons expel waves into the environment, their wave-number (frequency) and thus total wave-energy are reduced.

Quantization is Electronic, not Luminal

- 4. Electronic Wave-Energy Exchange with the Environment is Quantized: Other physical parameters of the wave-quanta that electrons absorb (amplitude, length, width) are fixed by the electron's wave-structure. Only the frequency-wavelength is variable and determines the energy of the quantum (E = hf).
- 5. Planck's Constant, *h*, is an Electron-Structure Constant: It describes electrons ($m_e = \frac{2R_{\infty}h}{c\alpha}$) and the quanta they emit and absorb. It is applicable to all phenomena and "particles" involving electrons and positrons. It does **not** describe free EM radiation.
- 6. Free EM Waves are not Quantized: A quantum of light is emitted as a distinct wave-packet, but upon emission begins to spread in space like all waves (Huygens-Fresnel principle). It ceases to exist qua quantum.

Directional Emission

- 7. Quanta are Emitted Directionally: As in a beam, not with spherical symmetry in all directions. Upon emission of a quantum, an electron recoils in the opposite direction. Individual electronic emissions therefore do not obey the inverse square law.
- 8. Spatial Spreading is Proportional to Wavelength: The shorter the wavelength, the less the spreading of the wave-packet after emission. At very short wavelengths, the emitted quantum may appear to not spread at all over short distances.

Background EM Radiation

- 9. Energetic Background Radiation: In any space, there is significant EM waveenergy of all frequencies from all near and distant sources (man-made, thermal, radioactive, solar, Cosmic, etc.). This radiant energy creates a highly energetic EM background (quantum fluctuations, the "mode").
- 10. Wave Interference is not Destructive: the amplitudes of innumerable waves from all sources at all distances are superimposed at any given point in space. Wave energy is not destroyed. The EM background is more energetic than previously assumed (quantum fluctuations).
- 11. Electrons are Coupled to the Background Radiation and other Electrons: An electron cannot exclude background waves. Its waves are constantly superpositioning with background waves and the waves of other electrons. This coupling induces both quantum absorptions and emissions.

Need for Statistical Method

- 12. The Absorbed Quantum is the Product of Superpositioning: Its energy does not usually come from the known source only, but from the superposition of source and background waves.
- 13. No Independent Knowledge of Emitters: In any laboratory setup, the location, timing, number, direction and spread of emitted quanta are unknown. Statements about emissions are only inferences from detection events.
- 14. Statistical Prediction: Since the quantum emissions in the source cannot be known, nor the background radiation or the state of the receiving electrons, we can only make statistical predictions concerning where and when detection events occur.

Answers to No Time Lag Objection

- Unknown Emitters: We cannot "see" emissions, only detections. A detected quantum absorption may be the product of many simultaneous quantum emissions in the source.
- Directional Emission: There is little diffractive spreading at shorter wavelengths, therefore the inverse-square law does not apply; a larger percentage of any single emitted quantum's wave-energy can impinge upon another electron.
- Electron Size: The electron is as large as its EM influence in space, so its reaction cross-section is larger than generally assumed.
- Background EM: The wave-energy that the electron absorbs does not come from the source alone, but from the superpositioning of source and background waves.

Demonstration of Wave-Principles

- In a low-light experimental setup, photoelectric detectors registered "dark" counts even when the source was not operating (energetic background EM radiation).
- The "photons" from a source were then filtered to the intensity of one-tenth "photon" (waves can be filtered, not particles).
- This sub-photonic EM wave-energy is sufficient, even at a distance of one meter, to produce additional photomultiplier counts (directional emission of the wave-packet plus superposition of source waves and background waves.)

Bennett, C., Brassard, G., Ekert, A., "Quantum Cryptography", Scientific American, 267, No. 4, 31 (1992)

An electron bound to an isolated hydrogen atom was detected, by its scattering of light, at a distance of several centimeters.

Aharanov, Y., quoted in "Quantum Philosophy", Sci. Am., 267, No. 1, 100 (1992)

 Experimentalist Carver Meade asserts that electrons are waves that expand to fit whatever container they are in. He claims that it is easy to make an electron that's 10 feet across.
Mead, C., Interview, The American Spectator, 70-75, Sept/Oct (2001)

Mead, C., Collective Electrodynamics, MIT Press (2000)

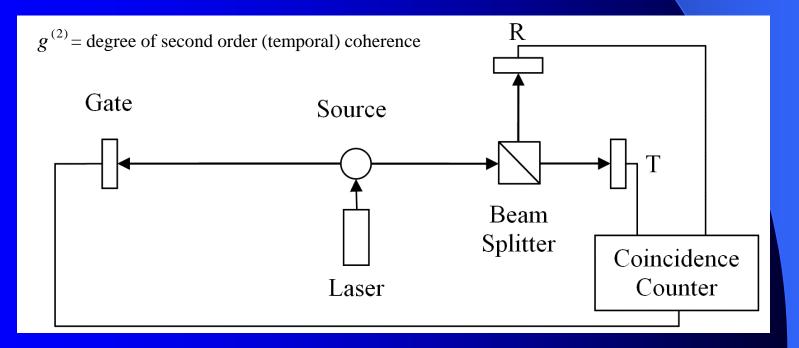
Anti-Correlation Experiments

Argument for Photon:

- 1. According to classical theory, a 50/50 BS sends equal wave amplitudes to R and T.
- 2. Therefore detections at R and T **should** be **correlated—both or none**.
- 3. Classical predication: $g^{(2)}(0) \ge 1$ Actual: $g^{(2)}(0) << 1$

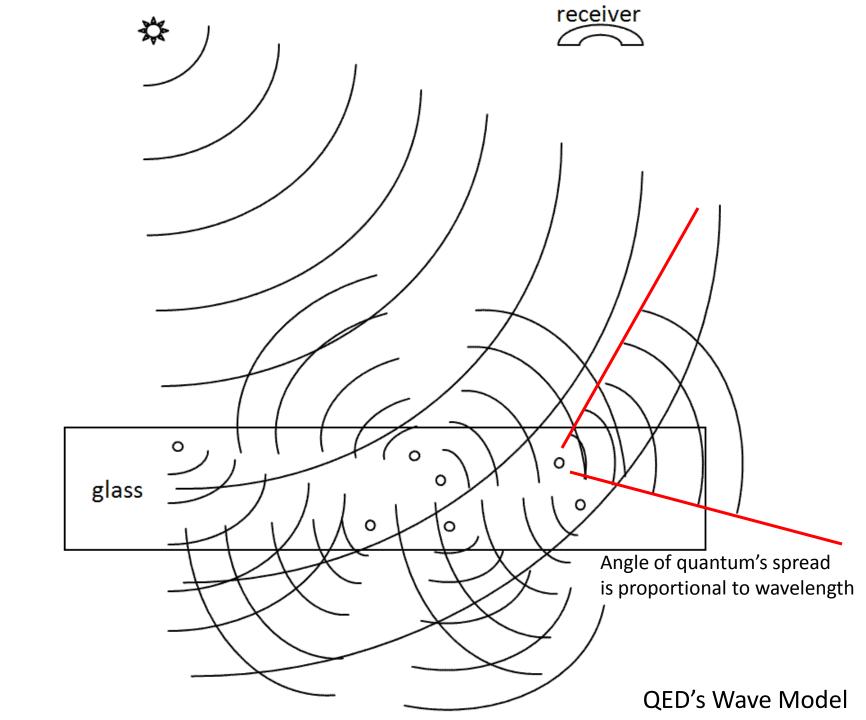
$$g^{(2)}(0) = \frac{N_{GTR}N_G}{N_{GT}N_{GR}}$$

- 4. Finding: In low intensity experiments, $N_{GTR}N_G << N_{GT}N_{GR}$
- 5. Conclusion: Light is particles; photon goes one way or the other at the BS. Thorn, J., Am. J. Phys, 72 (9) (2004)



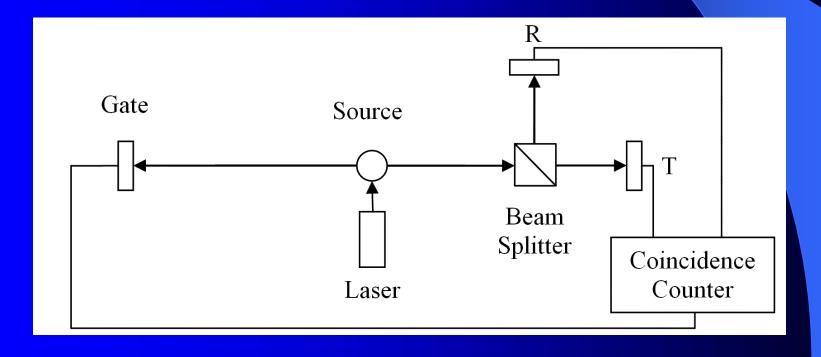
QED Wave-Theory Explanation

- 1. There are **no flying photons**, only spreading light wave **amplitudes** from an **unknown number** of source emissions and scatterings.
- 2. No. of emissions > No. of detections due to amplitude shrinking and scattering.
- 3. Non-Classical: Low intensity \rightarrow "graininess". Individual absorption/emission events dominate \rightarrow greater deviation from "classical" predictions
- 4. When there are only a **few** absorption/emission events in the BS, amplitude partitioning will **vary** from **50/50**; can even be **0/100**!
- 5. GRT coincidences will be rare : The BS scatters light in all directions, so if the source sends equal amplitudes towards the Gate and BS, just sufficient to produce a detection at the Gate, R and T will receive much smaller amplitudes.
- GR or GT coincidences will occur: sufficient amplitude at R or T requires some combination of multiple-emission output, uneven R-L down-converter output, uneven beam splitting, and a contribution from random background radiation.



Impossible, Miraculous Photons!

- 1. No flying photon has ever observed. It is a poor inference from detections.
- 2. Impossible knowledge: The number, direction, spreading, and scattering of quantum emissions from down-converter crystal and BS cannot be known.
- 3. Miraculous photons: supposed single flying photons are magically guided either straight through a BS or reflected intact, are "focused" by lenses, etc. How can light pass through mm's of glass (10²² atoms/cm³) and have no interaction with any electrons?



QED Reality: Light waves are scattered by air and by electrons throughout the glass!

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|---|----------------|------------------|
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| Superpositioning (interference) | Yes | Νο |
| Spreading, Diffraction | Yes | Νο |
| Continuous spectrum (gamma to radio waves) | Yes | Νο |
| Laser | Yes† | No* |
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| Photoelectric Effect | Yes† | No* |
| Compton Effect | Yes† | No* |
| Anti-correlation and other photon experiments | Yes† | No* |
| Quantum Electrodynamics | Yes† | No* |

* Requires <u>wave modeling</u> to predict "where the photon goes"
† Requires <u>quantization</u> of electronic absorption-emission of light, plus...

Light is a Wave

- The flying photon theory is **incompatible** with the evidence. (0% probability)
- Radio waves, x-rays, light, etc. are all waves in a medium: they differ only in their frequency, degree of spreading, how they are generated, and how they interact with matter.
- No Quantum/Classical schism: There is one electrodynamics that deals with both the quantized electronic absorption/emission of light and the nonquantized interactions.
- This wave theory of light and electrons is consistent with QED and eliminates all paradoxes.
- Nature is not absurd; it always makes sense when our theories conform to the facts.